

The aim of this dictionary is to present the terminology used in analytical chemistry. It comprises mutually correlated terminology from the whole range of analytical chemistry with special reference to newly developed branches and techniques. The dictionary is based on the terminological recommendations of the Analytical Section of the International Union of Pure and Applied Chemistry, collected in the Compendium of Analytical Nomenclature, and the papers published currently in Pure and Applied Chemistry. The present literature on analytical chemistry, comprising current publications and papers, textbooks, monographs and lexicons also served as sources for the preparation of the dictionary. The International System of Units (SI) was also taken into account. As a result, new concepts have been introduced into the dictionary, such as the amount of a substance and its unit - the mole, the elementary entity, the molar mass, and the equivalence factor. Much care has been taken to ensure the proper application of the above terms, especially in the definitions of various physicochemical quantities, laws, equations, constants and coefficients. Slow adaptation by chemists to new terms justifies the need to include in the dictionary terms which are not recommended by the International System. Their definitions are provided with appropriate remarks or indications regarding their correctness. The dictionary covers the following methods of analysis: qualitative, quantitative, elemental, optical, chromatographic, radiochemical, X-ray, electrochemical, spectrophotometric, and surface analysis. Terms covering laboratory equipment, statistics and error estimation, optimization of analytical methods, and design of experiments are also included. Scientists, researchers, engineers, students and all those who are concerned with modern analytical chemistry and scientific literature will find this dictionary invaluable.

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